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10/11/2023



Fluorine plays almost no role in biosphere

There are about 4700 naturally occurring organohalogen compounds



G. W. Gribble, J. Chem. Educ. 2004, 81, 1441; G. W. Gribble, Naturally Occurring Organohalogen Compounds-A Comprehensive Update 2010, 1; M. C. Walker, M. C. Y. Chang, Chem. Soc. Rev. 2014, 43, 6527; L. Wang, X. Zhou, M. Fredimoses, S. Liao, Y. Liu, RSC Advances 2014, 4, 57350. K. K. J. Chan, D. O'Hagan, in Methods Enzymol., 516, Academic Press, 2012, 219.

Fluorine is commonly used in bioactive molecules

To date, 20% of drugs and 30% of agrochemicals contain fluorine.



C. Ni, J. Hu, Synlett 2011, 770-782, G. K. S. Prakash, F. Wang, Chimica Oggi, 2012, 30, No. 5.

Fluorine-containing moieties can be used to mimic other functional groups

substituent	electronegativity	space filling model
$-CF_3$ (trifluoromethyl)	3.46	
–CF ₂ H (diifluoromethyl)	3.00	
–SH (thiol)	2.32	
–OH (hydroxyl)	3.51	
–CH ₃ (methyl)	2.27	
HO H ₂ N CO ₂ H	$\Rightarrow \qquad \stackrel{HF_{2}C}{\underset{H_{2}N}{\overset{C}}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}}{\overset{C}{\overset{C}}{\overset{C}{\overset{C}{\overset{C}{\overset{C}}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}}}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{\overset{C}{{}}}{\overset{C}{\overset{C}}}{\overset{{C}}{\overset{{C}}{\overset{C}}}{\overset{{C}}{\overset{{C}}{\overset{{C}}}{\overset{{C}}}{\overset{{C}}}{\overset{{C}}}{\overset{{C}}{{}}}{\overset{{C}}{{}}}{\overset{{C}}{{}}}{\overset{{C}}{{}}}{\overset{{C}}{{}}}{{}}{\overset{{C}}{{}}}{{}}}{{}}}{{}}}{{}}}{{}}}{{}}}{{}}}}$	CAREERSCO2H
Tyrosine naturally occurring amino acid)	Fluorinated tyrosir (fluorine-containing a	ne mimic mino acid)

E/R/TEAM

• Goals

- -Compute the physicochemical properties of fluorine-substituted amino acids using Density Functional Theory (DFT)
- -Compare these simulations to experimental results obtained in the lab
- —Create a model to predict the properties and interactions of these molecules



• Timeframe

-Start date: October 1st, 2023 -End date: April 1st, 2024



- What I hope to learn
 - -Become familiar with the HPC system
 - —Expand knowledge of Density Functional Theory, specifically within key molecules of a biological system
 - -Gain insights into how fluorine can function in an amino acid to confer desired properties



Goals for Next Month

 Finish running simulations on basic models to become familiar with the calculations and analysis

- Begin performing simulations and calculations on fluorinated amino acids
- -Begin obtaining experimental data for comparison

Help needed (if any)
–N/A

